Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

2,4,6-Trimethylpyridinium nitrate

Shahzad Sharif,[a] Mehmet Akkurt,[b]* Islam Ullah Khan,[a]* Abdul Rauf[a] and Irfana Mariam[a]

[a]Materials Chemistry Laboratory, Department of Chemistry, Government College University, Lahore 54000, Pakistan, and [b]Department of Physics, Faculty of Sciences, Erciyes University, 38039 Kayseri, Turkey
Correspondence e-mail: akkurt@erciyes.edu.tr, iukhan.gcu@gmail.com

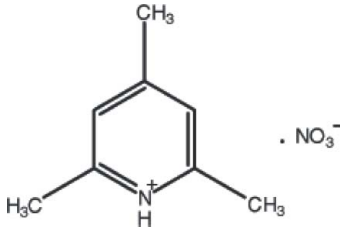
Received 2 August 2010; accepted 13 August 2010

Key indicators: single-crystal X-ray study; T = 296 K; mean σ(C–C) = 0.003 Å; R factor = 0.047; wR factor = 0.149; data-to-parameter ratio = 11.2.

In the title compound, $C_8H_{12}N^+\cdot NO_3^-$, the cation lies on a mirror plane and the N and one C atom lie on a twofold axis. In the crystal, the anions and cations are linked by N—H···O interactions along the b axis and a short N—O···π contact [3.2899 (5) Å] also occurs.

Related literature

For the use of *sym*-collidine and its derivatives, see: Brunel & Rousseau (1995); Homsi & Rousseau (1998); Rousseau & Robin (1997); Simonot & Rousseau (1994); Syper *et al.* (1980); Yamamoto *et al.* (1992). For structural properties of the related compound, 2,4,6-collidine, see: Bond & Davies (2001).



Experimental

Crystal data

$C_8H_{12}N^+\cdot NO_3^-$
M_r = 184.20
Orthorhombic, *Cmcm*
a = 9.328 (1) Å
b = 15.1327 (13) Å
c = 6.4967 (7) Å
V = 917.06 (16) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.10 mm^{-1}
T = 296 K
0.28 × 0.16 × 0.07 mm

Data collection

Bruker APEXII CCD diffractometer
1839 measured reflections
648 independent reflections
410 reflections with $I > 2\sigma(I)$
R_{int} = 0.030

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.047
$wR(F^2)$ = 0.149
S = 1.00
648 reflections
58 parameters
8 restraints
All H-atom parameters refined
$\Delta\rho_{max}$ = 0.16 e Å^{-3}
$\Delta\rho_{min}$ = −0.19 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1···O2^{i}	0.875 (18)	2.331 (16)	3.139 (3)	153.7 (2)
N1—H1···O2^{ii}	0.875 (18)	2.331 (16)	3.139 (3)	153.7 (2)

Symmetry codes: (i) $x+\frac{1}{2}, y-\frac{1}{2}, z$; (ii) $-x+\frac{3}{2}, y-\frac{1}{2}, -z+\frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON* (Spek, 2009).

The authors are grateful to the Higher Education Commission of Pakistan for financial support to purchase the diffractometer.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BX2300).

References

Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* **32**, 115–119.
Bond, A. D. & Davies, J. E. (2001). *Acta Cryst.* E**57**, o1141–o1142.
Bruker (2007). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Brunel, Y. & Rousseau, G. (1995). *Tetrahedron Lett.* **36**, 8217–8220.
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
Homsi, F. & Rousseau, G. (1998). *J. Org. Chem.* **63**, 5255–5258.
Rousseau, G. & Robin, S. (1997). *Tetrahedron Lett.* **38**, 2467–2470.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Simonot, B. & Rousseau, G. (1994). *J. Org. Chem.* **59**, 5912–5919.
Spek, A. L. (2009). *Acta Cryst.* D**65**, 148–155.
Syper, L., Kloc, K. & Młochowski, J. (1980). *Tetrahedron*, **36**, 123–129.
Yamamoto, K., Shimizu, M., Yamada, S., Iwata, S. & Hoshino, O. (1992). *J. Org. Chem.* **57**, 33–39.

supplementary materials

Acta Cryst. (2010). E**66**, o2362 [**doi:10.1107/S1600536810032629**]

2,4,6-Trimethylpyridinium nitrate

S. Sharif, M. Akkurt, I. U. Khan, A. Rauf and I. Mariam

Comment

Sym-collidine and its derivatives are extensively used in organic synthesis (Syper *et al.*, 1980; Rousseau *et al.*, 1997). Bis(2,4,6-trimethylpyridine)iodine(I) and -bromine(I) hexafluorophosphate have been used for specific electrophilic halogenations (Homsi *et al.*, 1998; Simonot *et al.*, 1994; Brunel *et al.*, 1995). It is also used in the synthesis of vitamin D (Yamamoto *et al.*, 1992). Here in we reported the crystal structure of collidinium nitrate.

In the title compound (I), (Fig. 1),the cation lies on a mirror plane and the N and one C atoms lies on two-fold axis. The anions and cations are linked by N—H···O interactions along the *b* axis. The bond distances and angles in (I) agree with those reported in a similar compound 2,4,6-collidine (Bond & Davies, 2001).

The anions and cations of (I) are linked by N—H···O interactions along the *b* axis (Table 1, Fig. 2). In the crystal structure, the O1 atom in the nitrate anion generates the N—O···π interactions [N2—O1···*Cg*1^{iii} = 3.2899 (5) Å and N2—O1···*Cg*1^{iv} = 3.2899 (5) Å; symmetry codes: (iii) -1/2 + *x*, 1/2 - *y*, -*z*; (iv) -1/2 + *x*, 1/2 - *y*, 1-*z*. *Cg*1 is a centroid of the aromatic pyridine ring] between two pyridine rings as a sandwich to establish the packing.

Experimental

To 2 ml of trimethyl pyridine, concentrated nitric acid (2 ml) was added drop wise. The mixture was refluxed for an hour, filtered. Within half an hour needle like crystals of titled compound appeared, suitable for *x*-ray crystallography.

Refinement

All H atoms were found on the difference map and refined with the distance restraints of N—H = 0.875 (18) Å and C—H = 0.93 (2) - 0.96 (4) Å. Their displacement parameters were constrained to ride on their parent atoms [U_{iso}(H) = 1.5U_{eq}(C) for methyl H atoms and 1.2U_{eq}(C,N) for other atoms].

Figures

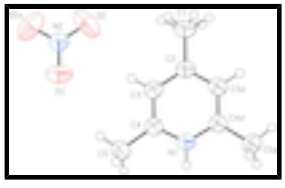


Fig. 1. A view of the title molecule. Displacement ellipsoids are drawn at the 50% probability level. [Symmetry codes: (*a*) 1-*x*, *y*, 1/2-*z*; (*d*) 2-*x*, *y*, 1/2-*z*.]

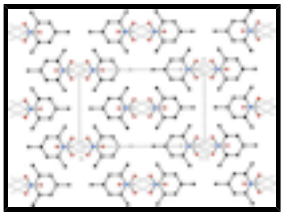


Fig. 2. A packing diagram of the title molecule showing the N—H···O interactions, down the *c* axis. All hydrogen atoms not involved in hydrogen bonding have been omitted for clarity.

2,4,6-Trimethylpyridinium nitrate

Crystal data

$C_8H_{12}N^+·NO_3^-$	$F(000) = 392$
$M_r = 184.20$	$D_x = 1.334$ Mg m^{-3}
Orthorhombic, *Cmcm*	Mo *K*α radiation, λ = 0.71073 Å
Hall symbol: -C 2c 2	Cell parameters from 494 reflections
$a = 9.328$ (1) Å	θ = 4.1–23.2°
$b = 15.1327$ (13) Å	μ = 0.10 mm^{-1}
$c = 6.4967$ (7) Å	$T = 296$ K
$V = 917.06$ (16) Å^3	Needle, colourless
$Z = 4$	0.28 × 0.16 × 0.07 mm

Data collection

Bruker APEXII CCD diffractometer	410 reflections with $I > 2\sigma(I)$
Radiation source: sealed tube	$R_{int} = 0.030$
graphite	$\theta_{max} = 28.3°$, $\theta_{min} = 4.1°$
φ and ω scans	$h = -11 \rightarrow 12$
1839 measured reflections	$k = -19 \rightarrow 20$
648 independent reflections	$l = -8 \rightarrow 4$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.047$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.149$	All H-atom parameters refined
$S = 1.00$	$w = 1/[\sigma^2(F_o^2) + (0.0764P)^2 + 0.2375P]$ where $P = (F_o^2 + 2F_c^2)/3$
648 reflections	$(\Delta/\sigma)_{max} < 0.001$
58 parameters	$\Delta\rho_{max} = 0.16$ e Å^{-3}
8 restraints	$\Delta\rho_{min} = -0.19$ e Å^{-3}

Special details

Geometry. Bond distances, angles *etc*. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted *R*-factors *wR* and all goodnesses of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating -*R*-factor-obs *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}	Occ. (<1)
N1	1.00000	0.13163 (15)	0.25000	0.0411 (8)	
C1	1.00000	0.4105 (2)	0.25000	0.0604 (13)	
C2	1.00000	0.31135 (19)	0.25000	0.0444 (10)	
C3	0.8725 (2)	0.26452 (14)	0.25000	0.0454 (7)	
C4	0.8728 (2)	0.17377 (14)	0.25000	0.0422 (7)	
C5	0.7402 (3)	0.11973 (17)	0.25000	0.0577 (9)	
O1	0.50000	0.31452 (15)	0.25000	0.0754 (10)	
O2	0.6109 (2)	0.43578 (17)	0.25000	0.1128 (13)	
N2	0.50000	0.39433 (16)	0.25000	0.0469 (9)	
H1	1.00000	0.0738 (12)	0.25000	0.0560*	
H1A	1.095 (3)	0.435 (4)	0.25000	0.0700*	0.500
H1B	0.949 (3)	0.432 (2)	0.369 (4)	0.0700*	0.500
H3	0.7858 (19)	0.2953 (14)	0.25000	0.0560*	
H5A	0.661 (2)	0.1569 (14)	0.25000	0.0700*	
H5B	0.7366 (19)	0.0801 (10)	0.135 (3)	0.0700*	

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0464 (14)	0.0334 (12)	0.0436 (16)	0.0000	0.0000	0.0000
C1	0.065 (2)	0.0383 (16)	0.078 (3)	0.0000	0.0000	0.0000
C2	0.0526 (17)	0.0367 (14)	0.044 (2)	0.0000	0.0000	0.0000
C3	0.0451 (11)	0.0428 (12)	0.0483 (15)	0.0043 (9)	0.0000	0.0000
C4	0.0428 (11)	0.0432 (11)	0.0407 (14)	−0.0009 (9)	0.0000	0.0000
C5	0.0470 (13)	0.0491 (13)	0.077 (2)	−0.0054 (10)	0.0000	0.0000
O1	0.098 (2)	0.0401 (12)	0.088 (2)	0.0000	0.0000	0.0000
O2	0.0894 (16)	0.1011 (18)	0.148 (3)	−0.0507 (13)	0.0000	0.0000
N2	0.0526 (15)	0.0461 (15)	0.0420 (17)	0.0000	0.0000	0.0000

Geometric parameters (Å, °)

O1—N2	1.208 (3)	C1—H1A^{i}	0.96 (4)

O2—N2	1.210 (2)	C1—H1A	0.96 (4)
N1—C4	1.347 (2)	C1—H1B	0.96 (3)
N1—C4^{i}	1.347 (2)	C1—H1B^{ii}	0.96 (3)
N1—H1	0.875 (18)	C1—H1B^{i}	0.96 (3)
C1—C2	1.500 (4)	C1—H1B^{iii}	0.96 (3)
C2—C3^{i}	1.384 (2)	C3—H3	0.933 (19)
C2—C3	1.384 (2)	C5—H5B^{iii}	0.959 (18)
C3—C4	1.373 (3)	C5—H5A	0.93 (2)
C4—C5	1.483 (3)	C5—H5B	0.959 (18)
O2···N1^{iv}	3.139 (3)	H1···H5B^{i}	2.570 (18)
O2···C5^{v}	3.111 (4)	H1···H5B^{iii}	2.570 (18)
O2···N1^{v}	3.139 (3)	H1···H5B	2.570 (18)
O1···H3	2.682 (18)	H1···O2^{ix}	2.331 (16)
O1···H5A^{vi}	2.82 (2)	H1···N2^{ix}	2.716 (18)
O1···H3	2.682 (18)	H1···H5B^{ii}	2.570 (18)
O1···H5A	2.82 (2)	H1···O2^{xii}	2.331 (16)
O1···H3^{vi}	2.682 (18)	H1···O2^{xvi}	2.331 (16)
O1···H3^{vii}	2.682 (18)	H1···N2^{xvi}	2.716 (18)
O1···H5A^{vii}	2.82 (2)	H1···N2^{xiii}	2.716 (18)
O1···H5A	2.82 (2)	H1···N2^{xii}	2.716 (18)
O2···H1^{iv}	2.331 (16)	H1···O2^{xiii}	2.331 (16)
O2···H5B^{viii}	2.888 (19)	H1A···H3^{i}	2.39 (6)
O2···H3	2.68 (2)	H1A···O2^{i}	2.74 (3)
O2···H1A^{i}	2.74 (3)	H1A···O2^{ii}	2.74 (3)
O2···H3	2.68 (2)	H1A···H3^{i}	2.39 (6)
O2···H1A^{i}	2.74 (3)	H1B···H5B^{viii}	2.45 (3)
O2···H1^{v}	2.331 (16)	H3···O1	2.682 (18)
O2···H5B^{v}	2.711 (16)	H3···O2	2.68 (2)
N1···N2^{viii}	3.2720 (5)	H3···H5A	2.40 (3)
N1···O2^{ix}	3.139 (3)	H3···O1	2.682 (18)
N1···N2^{x}	3.2720 (5)	H3···O2	2.68 (2)
N1···N2^{xi}	3.2720 (5)	H3···H1A^{i}	2.39 (6)
N1···O2^{xii}	3.139 (3)	H3···O1	2.682 (18)
N1···O2^{xiii}	3.139 (3)	H3···O1	2.682 (18)
N1···N2^{xiv}	3.2720 (5)	H3···H1A^{i}	2.39 (6)
N1···N2^{xv}	3.2720 (5)	H5A···O1	2.82 (2)
N1···O2^{xvi}	3.139 (3)	H5A···O1	2.82 (2)
N1···N2^{xvii}	3.2720 (5)	H5A···H3	2.40 (3)
N1···N2^{xviii}	3.2720 (5)	H5A···O1	2.82 (2)
N1···N2^{xix}	3.2720 (5)	H5A···O1	2.82 (2)
N2···N1^{viii}	3.2720 (5)	H5B···O2^{x}	2.888 (19)

N2···N1^{x}	3.2720 (5)	H5B···H1B^{x}	2.45 (3)
N2···H1^{iv}	2.716 (18)	H5B···H1	2.570 (18)
N2···H1^{v}	2.716 (18)	H5B···O2^{xix}	2.888 (19)
C5···O2^{xiii}	3.111 (4)	H5B···O2^{xiii}	2.711 (16)
C5···O2^{xvi}	3.111 (4)	H5B···O2^{xvi}	2.711 (16)
C4—N1—C4^{i}	123.5 (2)	H1A^{i}—C1—H1B	54.2 (17)
C4—N1—H1	118.26 (12)	H1B—C1—H1B^{i}	141 (3)
C4^{i}—N1—H1	118.26 (12)	H1B—C1—H1B^{iii}	107 (2)
O2—N2—O2^{vi}	117.5 (3)	H1A—C1—H1A^{i}	135 (5)
O1—N2—O2^{vi}	121.23 (15)	H1A—C1—H1B^{i}	54.2 (17)
O1—N2—O2	121.23 (15)	H1A^{i}—C1—H1B^{iii}	54.2 (17)
C1—C2—C3	120.79 (13)	H1A^{i}—C1—H1B^{ii}	109 (2)
C1—C2—C3^{i}	120.79 (13)	H1B^{i}—C1—H1B^{iii}	59 (2)
C3—C2—C3^{i}	118.4 (2)	H1B^{i}—C1—H1B^{ii}	107 (2)
C2—C3—C4	120.67 (19)	H1B^{iii}—C1—H1B^{ii}	141 (3)
N1—C4—C5	118.3 (2)	C2—C1—H1A	113 (3)
C3—C4—C5	123.35 (19)	H1B—C1—H1B^{ii}	59 (2)
N1—C4—C3	118.37 (18)	H1A^{i}—C1—H1B^{i}	109 (2)
C2—C1—H1A^{i}	113 (3)	C2—C3—H3	119.3 (13)
C2—C1—H1B^{i}	109.7 (18)	C4—C3—H3	120.1 (13)
C2—C1—H1B^{ii}	109.7 (18)	C4—C5—H5B^{iii}	112.0 (11)
H1A—C1—H1B	109 (2)	H5A—C5—H5B^{iii}	110.6 (13)
C2—C1—H1B^{iii}	109.7 (18)	H5B—C5—H5B^{iii}	102.4 (15)
C2—C1—H1B	109.7 (18)	H5A—C5—H5B	110.6 (13)
H1A—C1—H1B^{iii}	109 (2)	C4—C5—H5A	109.2 (13)
H1A—C1—H1B^{ii}	54.2 (17)	C4—C5—H5B	112.0 (11)
C1—C2—C3—C4	180.00	C2—C3—C4—C5	180.00
C2—C3—C4—N1	0.00		

Symmetry codes: (i) −*x*+2, *y*, −*z*+1/2; (ii) −*x*+2, *y*, *z*; (iii) *x*, *y*, −*z*+1/2; (iv) *x*−1/2, *y*+1/2, *z*; (v) −*x*+3/2, *y*+1/2, −*z*+1/2; (vi) −*x*+1, *y*, −*z*+1/2; (vii) −*x*+1, *y*, *z*; (viii) −*x*+3/2, −*y*+1/2, *z*+1/2; (ix) *x*+1/2, *y*−1/2, *z*; (x) −*x*+3/2, −*y*+1/2, *z*−1/2; (xi) −*x*+3/2, −*y*+1/2, −*z*+1; (xii) *x*+1/2, *y*−1/2, −*z*+1/2; (xiii) −*x*+3/2, *y*−1/2, *z*; (xiv) *x*+1/2, −*y*+1/2, *z*−1/2; (xv) *x*+1/2, −*y*+1/2, *z*+1/2; (xvi) −*x*+3/2, *y*−1/2, −*z*+1/2; (xvii) *x*+1/2, −*y*+1/2, −*z*; (xviii) *x*+1/2, −*y*+1/2, −*z*+1; (xix) −*x*+3/2, −*y*+1/2, −*z*.

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N1—H1···O2^{ix}	0.875 (18)	2.331 (16)	3.139 (3)	153.7 (2)
N1—H1···O2^{xvi}	0.875 (18)	2.331 (16)	3.139 (3)	153.7 (2)

Symmetry codes: (ix) *x*+1/2, *y*−1/2, *z*; (xvi) −*x*+3/2, *y*−1/2, −*z*+1/2.

Fig. 1

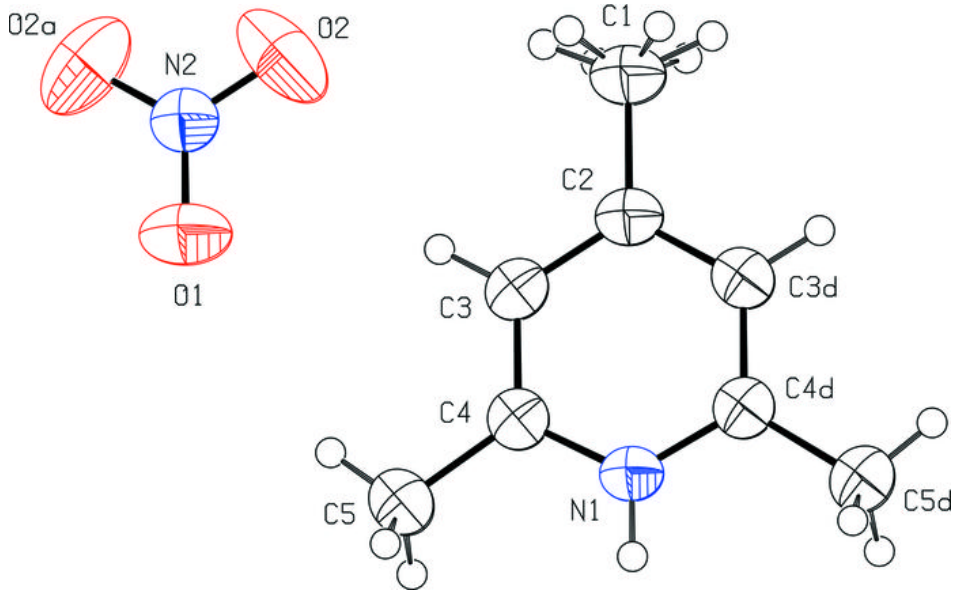


Fig. 2